## BRIEF COMMUNICATION

# Strong Correlation Between Statistical Transmembrane Tendency and Experimental Hydrophobicity Scales for Identification of Transmembrane Helices

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Abstract Direct physical chemistry measurements of the hydrophobicity of amino acids or their derivatives have often been used to estimate the propensity of amino acids to participate in transmembrane helices. In this short note, it is found that there is a very high degree of correlation  $(r = 0.944 - 0.965)$  between an average physical chemistry hydrophobicity scale (an average of scales derived, e.g., from the solubility of amino acid derivatives in organic solvents versus water or their binding to hydrophobic particles) and the statistically based transmembrane tendency scale (derived from the relative abundance of residues in known transmembrane and soluble protein sequences (Zhao and London, Protein Sci 15:1987–2001, 2006)). This correlation indicates that, other than hydrophobicity, amino acid properties/interactions that promote or inhibit transmembrane helix formation in a specific membrane protein largely cancel out when averaged over all transmembrane sequences. In other words, other than hydrophobicity, there are no properties of a specific amino acid residue within a hydrophobic segment that have a strong systematic effect upon transmembrane helix formation independent of the remainder of the sequence in that hydrophobic segment. However, proline is an exception to this rule.

Keywords Transmembrane protein · Hydrophobic alpha helices  $\cdot$  Hydrophobicity  $\cdot$ Biological hydrophobicity · Transmembrane tendency

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### Introduction

Prediction of transmembrane (TM) helices is a classical problem in biochemistry/molecular biology. The utility of scales that predict the TM segments of a protein based on their amino acid sequence is obvious. Considerable effort has been expended in trying to define the most accurate method to derive such scales. Two general approaches have been investigated. In the physical chemistry approach, the partitioning of amino acid-containing molecules between an aqueous and a hydrophobic environment is measured (Kyte and Doolittle [1982;](#page-3-0) Roseman [1988](#page-3-0); Wolfenden et al. [1981](#page-3-0)). The hydrophobic environment can be a solvent or a lipid bilayer (Kessel et al. [2003;](#page-3-0) Kyte and Doolittle [1982](#page-3-0); Wimley and White [1996\)](#page-3-0). A biological analogue to this approach has been recently developed (Hessa et al. [2005,](#page-3-0) [2007](#page-3-0), [2009;](#page-3-0) Xie et al. [2007\)](#page-3-0). In this elegant method the ability of simple hydrophobic sequences in a chimeric protein to form a TM structure is used to evaluate the effect of amino acid sequence upon TM insertion (Hessa et al. [2005](#page-3-0), [2007](#page-3-0), [2009](#page-3-0); Xie et al. [2007](#page-3-0)). The second approach is a statistical one in which the abundance of amino acids within TM helices and soluble sequences is compared. This can be used to develop a simple scale in which each amino acid is assigned a single value representing its propensity to be in a TM sequence (Zhao and London [2006\)](#page-3-0) or as part of a more sophisticated analysis (i.e., hidden Markov methods) in which all the information in TM and surrounding sequences is used to predict TM helices (Kahsay et al. [2005](#page-3-0)). We previously derived a statistical type of TM tendency scale based on the amino acid composition of sequences in databases of soluble and TM sequences from both eukaryotes and prokaryotes (Zhao and London [2006](#page-3-0)). It was demonstrated that the TM tendency scale represents a scale at the theoretical limit of accuracy for a statistical

scale in which the abundance of amino acids in soluble and TM proteins is used to define a single average value for the propensity of each type of amino acid to occur in a TM helix. We also found a strong correlation between TM tendency and an experimental scale developed using natural membranes and based on what sequences processed by the translocon will form TM helices (Zhao and London [2006\)](#page-3-0). This indicates that the TM tendency scale effectively captures the average behavior of natural sequences. In this brief note we show that an average physical chemistry-based hydrophobicity scale is highly correlated to TM tendency and consider the implications of this correlation.

### Results and Discussion

Table 1 shows an average physical chemistry hydrophobicity scale (AvgH scale) we derived by averaging values from 15 different published scales that we were able to identify in the literature (Abraham and Leo [1987;](#page-2-0) Black and Mould [1991;](#page-2-0) Browne et al. [1982;](#page-2-0) Bull and Breese [1974;](#page-2-0) Cowan and Whittaker [1990](#page-2-0); Deber et al. [2001](#page-2-0); Jayasinghe et al. [2001](#page-3-0); Kessel et al. [2003;](#page-3-0) Kyte and Doolittle [1982](#page-3-0); Meek [1980](#page-3-0); Parker et al. [1986;](#page-3-0) Roseman [1988](#page-3-0); Wilson et al. [1981](#page-3-0); Wolfenden et al. [1981](#page-3-0)). To avoid introducing any bias, we used all the scales we could find, rather than choose a specific subset of scales for detailed analysis. The scales were normalized to have an equal span

> AvgH Phe 8.88 8.85 Ile 8.38 Trp 8.26 7.19 6.63 6.58 5.84 5.46 5.15 4.55 4.28 3.82 3.47 Gln 3.18 2.80 2.24 1.74 Glu 1.71



values o

of values between the most hydrophobic and most hydrophilic amino acid residues and, where necessary, reversed in sign so that larger values correspond to higher hydrophobicity. The resulting AvgH scale shows a close correlation  $(r = 0.944)$  to the TM tendency scale (Zhao and London [2006](#page-3-0)). Notice that this degree of correlation is generally higher than that of any individual experimental hydrophobicity scales with the TM tendency scale or their correlation with each other (Table [2](#page-2-0)). The good correlation between hydrophobicity (AvgH) and TM tendency indicates that they measure a very similar set of overall amino acid properties. The correlation is even better  $(r = 0.965)$ if Pro, which is an outlier that appears more hydrophobic in the physical chemistry scales than in the TM tendency scale (Fig. [1](#page-2-0)), is ignored. The fact that Pro has a lower TM tendency than expected based upon its physical chemical hydrophobicity may partly reflect its inability to form a proper backbone hydrogen bond in the context of a helix. This property would not be detected in hydrophobicity measurements using isolated Pro or other non-helix-forming model compounds containing Pro.

The strong correlation between the AvgH and TM scales leads to the conclusion that (except for Pro) factors other than hydrophobicity have little systematic effect on whether a sequence forms a TM helix. This is not to say that specific polar interactions between residues do not have a crucial importance in TM helix formation in individual membrane proteins. Instead, it implies that there are no properties of amino acids other than hydrophobicity that strongly promote or interfere with TM helix formation without regard to what other residues are present in the TM helix.

This conclusion is based upon the use of AvgH, an average experimental hydrophobicity scale; and it is reasonable to ask: Why should this AvgH scale give a more accurate result than the individual experimental scales from which it is derived? There are various physical chemical methods to assess hydrophobicity (solvent partition, HPLC retention times, surface tension, vapor pressure, and computational methods) and various amino acid derivatives that can be used for such measurements. We made the assumption that most such scales derived from these approaches would have peculiarities specific to the system used to measure hydrophobicity. For example, if the solubility of some amino acid derivatives in different hydrophobic solvents/environments is measured, then the specific properties of that solvent/hydrophobic environment (polarity, hydrogen bonding capability) and the way it interacts with the specific amino acid derivative chosen could influence the scale in some systematic fashion. This is supported by the data in Table [2](#page-2-0), which show that the correlation between different experimental hydrophobicity scales is often poor. However, if these peculiarities are

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The scales shown are from the following references: TM tendency (Zhao and London [2006\)](#page-3-0); KD (Kyte and Doolittle [1982](#page-3-0)); CW (Cowan and Whittaker 1990); AL (Abraham and Leo 1987); BB (Bull and Breese 1974); Deber et al. (2001); Roseman ([1988\)](#page-3-0); Wolfenden et al. [\(1981](#page-3-0)); Wilson et al. [\(1981](#page-3-0)); Browne et al., TFA or HFBA (Browne et al. 1982); BM (Black and Mould 1991); Parker et al. ([1986](#page-3-0)); Meek [\(1980](#page-3-0)); Jayasinghe et al. ([2001\)](#page-3-0); and Kessel et al. [\(2003](#page-3-0))



Fig. 1 Correlation between TM tendency and average hydrophobicity values. Zero values are arbitrary for these scales. Notice that the TM tendency of Pro is significantly lower than that expected based on its physical chemical hydrophobicity. See text for details

specific to each individual scale, they should tend to cancel out in the average of different scales. As the comparison to TM tendency shows, this assumption is likely to be valid.

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